UTKAL GOURAV MADHUSUDAN INSTITUTE OF TECHNOLOGY, RAYAGADA Academic Lesson Plan for 2nd Semester - 2025 (Summer)

Name of the teaching faculty: Miss Dasmanti sabar,

Guest faculty (Chemistry)

Discipline : Common (Civil/Electrical/E&TC/Mechanical)

Dept. : Department of Mathematics & Science

Semester : 2nd

Subject : Pr 5 : Applied Chemistry Laboratory

No of Periods per week: 4 Total Periods: 30, End semester Exam.: 25 Marks, Sessional: 25Marks,

Total Marks: 50 Marks

Week	Period	Experiment	Topics to be covered
1st	1 st		Introduction to chemistry lab, about safety measures, about maintenance of practical records.
	2 nd		Introduction to the students about use of different lab equipments and how to handle them safely.
	3rd	Exp.1	Preparation of standard solution of oxalic acid and potassium permanganate.
	4 th	Exp. 2	Acid base titration of oxalic acid against sodium hydroxide using phenolphthalein indicator.
2nd	1 st	Exp.1	Checking observation record.
	2 nd	Exp. 3	Standardization of KMnO ₄ solution against oxalic acid.
	3rd	Exp.3	Determine the percentage of iron present in the given hematite ore by KMnO ₄ solution.
	4 th	Exp.2	Expt. Conducted by the Students
	1 st	Exp.2	Checking of rough practical record
0.1	2 nd	Exp.3	Expt. Conducted by the Students expt.3
3rd	3 rd	Exp.3	Expt. Conducted by the Students expt.3
	4 th	Exp.3	Checking of rough practical record
4th	1 st	Exp.1	Checking of practical records and discussion of viva questions of expt. 1
	2 nd	Exp.2	Checking of practical records and discussion of viva questions of expt. 2
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	3rd	Exp.3	Checking of practical records and discussion of viva questions of expt. 3.
	4 th	Exp.4	Introduction of iodometric estimation of copper in the copper pyrite ore.
	1st	Exp.4	Demonstration of iodometric estimation.
5th	2 nd	Exp.4	Checking of rough practical record
	3 rd	Exp.4	Expt. Conducted by the Students.
	4th	Exp.4	Expt. Conducted by the Students.
	1 st	Exp.4	Checking of practical records and discussion of viva questions of expt. 4.
6th	2 nd	Exp.4	Checking of practical records and discussion of viva questions of expt. 4.
	3rd	Exp.5	Introduction and explanation of total acid number (TAN) of given oil.
	4 th	Exp.5	Explanation of procedure of exp.5 and discussion.
ļ	1 st	Exp.5	Demonstrate of the expt. 5
	2 nd	Exp.5	Expt. Conducted by student
7th	3rd	Exp.5	Checking of rough practical record
	4 th	Exp.5	Checking of practical records and discussion of viva question of expt.5
8th	1 st	Exp.6	Introduction of determining the conductivity of given water sample.
	2 nd	Exp.6	Explanation of experiment 6 procedure and discussion.
	3rd	Exp.6	Demonstrating expt.6
	4 th	Exp.6	Expt. Conducted by the Students
9th	1 st	Exp.6	Expt. Conducted by the Students
	2 nd	Exp.6	Checking of observation record.

4th Exp.7 Introduction of determination of iron content in given cement sample using colorimeter.		3 rd	Exp.6	Checking of practical records and discussion of viva questions of expt.6
10th 2nd Exp.7 Expt. Conducted by the Students		4 th	Exp.7	Introduction of determination of iron content
10th 2nd Exp.7 Expt. Conducted by the Students				
1st Exp.8 Demonstrating the working of redwood viscometer.		1 st	Exp.7	
Exp.7 Checking observation record.		2 nd	Exp.7	Expt. Conducted by the Students
Exp.7 Checking of practical records and discussion of viva questions of expt.7	10th		Exp.7	Checking observation record.
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Out.		2 nd	Exp.10	To verify the first law of electrolysis of copper
		3 rd	Exp.10	

	4 th	Exp.10	Demonstration of the expt.10 and explanation of the working principle.
	1 st	Exp.10	Explaining the procedure exp.10.
15th	2 nd	Exp.10	Checking of rough observation record
	3 rd	Exp.10	Checking of practical records and discussion of viva questions of expt.10.
	4 th	Exp.10	Checking of practical records and discussion of viva questions of expt.10.

Miss. Dasmanti Sabar Guest faculty (Chemistry), Dept. of Mathematics & Science, UGMIT, Rayagada